

CONTRIBUTIONS TO THE QR FORUM

What is diatomite?

Petra Zahajská^{a,b,*} , Sophie Opfergelt^c, Sherilyn C. Fritz^d, Johanna Stadmark^a, Daniel J. Conley^a

^aDepartment of Geology, Lund University, Sölvegatan 12, 22362 Lund, Sweden

^bInstitution of Palaeontology and Geology, Charles University, Albertov 6, 12832 Prague, Czechia

^cEarth and Life Institute, Université catholique de Louvain, 1348 Louvain-la-Neuve, Belgium

^dDepartment of Earth and Atmospheric Sciences, School of Biological Sciences, University of Nebraska, Lincoln, Nebraska 68588-0340 USA

*Corresponding author at: Department of Geology, Lund University, Sölvegatan 12, 22362 Lund, Sweden. E-mail address: petra.zahajska@geol.lu.se (P. Zahajská).

(RECEIVED August 19, 2019; ACCEPTED February 15, 2020)

Abstract

Different types of biogenic remains, ranging from siliceous algae to carbonate precipitates, accumulate in the sediments of lakes and other aquatic ecosystems. Unicellular algae called diatoms, which form a siliceous test or frustule, are an ecologically and biogeochemically important group of organisms in aquatic environments and are often preserved in lake or marine sediments. When diatoms accumulate in large numbers in sediments, the fossilized remains can form diatomite. In sedimentological literature, “diatomite” is defined as a friable, light-coloured, sedimentary rock with a diatom content of at least 50%, however, in the Quaternary science literature diatomite is commonly used as a description of a sediment type that contains a “large” quantity of diatom frustules without a precise description of diatom abundance. Here we pose the question: What is diatomite? What quantity of diatoms define a sediment as diatomite? Is it an uncompacted sediment or a compacted sediment? We provide a short overview of prior practices and suggest that sediment with more than 50% of sediment weight comprised of diatom SiO₂ and having high (>70%) porosity is diatomaceous ooze if unconsolidated and diatomite if consolidated. Greater burial depth and higher temperatures result in porosity loss and recrystallization into porcelanite, chert, and pure quartz.

Keywords: Diatomite; Sediment classification; Biogenic sediment; Diatomaceous sediment; Diatomaceous ooze

In many lakes worldwide, as well as in marine environments, diatoms are an important component of phytoplankton populations. Accumulations of diatoms are known from all aquatic environments including wetlands, lakes, and the marine environment (Clarke, 2003). In marine environments, accumulations of diatom-rich sediments extend back to the Late Cretaceous (Harwood et al., 2007), whereas the oldest diatom-rich lake sediments are from the Eocene (Flower et al., 2013), with large deposits found in the Miocene (Bradbury and Krebs, 1995). Massive accumulations of fossil diatom frustules have been observed in multiple lakes situated in silica-rich environments, especially in volcanic and

hydrothermally active areas, such as Yellowstone Lake, US (Theriot et al., 2006), Lake Myvatn, Iceland (Opfergelt et al., 2011), or Lake Challa, Tanzania/Kenya (Barker et al., 2013). In these settings, the high dissolved silicon concentrations promote the growth of diatoms (Wallace, 2003). However, high diatom concentrations in sediment also have been observed in lakes with no volcanic or hydrothermal influence, for example in lakes of Northern Sweden (Frings et al., 2014) or Lough Neagh, Ireland (Plunkett et al., 2004). In the oceanic environment, high biogenic silica accumulations occur in the equatorial Pacific Ocean, where diatoms grow in zones fed by continental siliceous dust and nutrients brought by upwelling. Similarly, cold-water regions, such as the productive Antarctic convergence zone, have sufficient nutrient and dissolved silicon supply that diatom-rich sediments are formed (Flower et al., 2013).

Cite this article: Zahajská, P., Opfergelt, S., Fritz, S. C., Stadmark, J., Conley, D. J. 2020. What is diatomite? *Quaternary Research* 96, 48–52. <https://doi.org/10.1017/qua.2020.14>

Conger (1942) hypothesized that the requirements for diatomaceous accumulation are to have (1) conditions favourable for diatom growth, and (2) a reduced accumulation of other sedimentary constituents that would dilute the concentration of the siliceous tests of diatoms. Diatom growth is dependent on many environmental factors, such as dissolved silicon availability, phosphorus and nitrogen availability, pH, salinity, and light (Battarbee et al., 2002). Diatom test preservation is sensitive to temperature and pH; biogenic silica dissolves faster with increasing pH (>8) and temperature (Alexander et al., 1954). Although diatom-rich sediments are found worldwide in various environments, of various ages and various settings, the terminology for classifying high accumulations of diatom frustules in sediment is not consistent in the literature, particularly in the Quaternary literature.

Sediments can be classified based on the sedimentary environment, sedimentary structures and processes, and sediment texture and composition, including sorting, shape of grains, grain size, and the ratio of matrix to grains. Grain size, for example, is used in various classification systems (e.g., Wentworth, 1922 or see summary in Pettijohn [1949]). This is appropriate for minerogenic sediments, but biogenic sediments contain more than just mineral grains. Biogenic sediments (also called bioclastic or organic) are composed of skeletal remains, shells, or tests from organisms composed of biogenic silica or calcium carbonate, or remains of soft organic material, mainly organic carbon. The classification of calcium carbonate sediments based on texture and the ratio of matrix to the abundance of grains is well described by Dunham (1962), and an effective general classification of lacustrine sediments was proposed by Schnurrenberger (2003), but it has not been widely used.

The first usage of the word diatomite dates to the nineteenth century from deep-sea deposits that were called diatom ooze, collected during the voyage of the HMS Challenger (Murray and Renard, 1891). Later, Conger (1942) described pure diatomaceous earth as material that reached a purity of 95 to 98% of diatom silica. According to Terzaghi et al. (1996), diatomaceous ooze should be used for loose unconsolidated sediment containing mainly diatoms. In colloquial nonscientific literature, diatomaceous earth is used as a name for both milled diatomite and for diatomaceous ooze, which creates ambiguity in the usage of diatomaceous earth as a definition. For this reason, we exclude diatomaceous earth from our proposed classification, which is intended for use in scientific literature rather than in the public domain.

Inglethorpe (1993) described the characteristics of diatomite as a unique combination of physical and chemical properties (high porosity, high permeability, small particle size, large surface area, low thermal conductivity, and chemical inertness) that make diatomite suitable for a wide range of industrial applications. Diatomite was defined as a “pale coloured, soft, light-weight rock composed principally of the silica microfossils” (Inglethorpe, 1993, p. 1). A diatomite of high SiO₂ purity (ranging from 80 to 99 wt% of biogenic SiO₂) is now commonly used in scientific research as a

reference material in isotope geochemistry for the measurement of stable silicon isotopes. One widely used standard (Reynolds et al., 2007) originates from the Lompoc quarry in California, more precisely from the Miocene strata of the Monterey Formation, which is well known for numerous lithological stages of siliceous deposits—diatomite, diatomaceous shales, diatomaceous mudstones, porcelanite, and cherts (Bramlette, 1946). The Lompoc area is well described by Bramlette (1946), including a description of the purity of diatomaceous deposits.

Various generalized classification systems for lacustrine and marine sediments have been proposed by Dean (1985), Mazzullo et al. (1988), Owen (2002), and Schnurrenberger (2003). Dean (1985) and Mazzullo et al. (1988) proposed that biogenic sediment be defined as sediment that contains at least 50% of biogenic material. In this approach, biogenic content is estimated visually or by point-counting on smear slides, which has been an efficient method for calcareous nanoplankton sediment classification. For siliceous nanoplankton (diatoms, radiolarians, and chrysophyte cysts), which are comparatively big and porous, the point-counting method can both overestimate (Dean et al., 1985) and underestimate (Conley, 1988) the percentage of biogenic silica. An alternative approach is to use the wt% biogenic silica relative to the dry sediment to classify unconsolidated biogenic silica-rich sediment using the methods of DeMaster (1981), Morlock and Froelich (1989), or Conley and Schelske (2001). However, the weak base extractions dissolve only biogenic silica that is classified as opal-A, and sediments that are more diagenetically altered containing highly ordered opal-CT require a much stronger base.

A detailed classification based on diatom SiO₂ content by Owen (2002) does not consider the stage of lithification caused by diagenetic processes. Diagenetic processes decrease the sediment porosity and alter the crystalline structure of SiO₂. For example, non-crystalline opal-A (such as the SiO₂ found in live siliceous organisms and their biogenic remains) is transformed into the disordered silica polymorph opal-CT (Rice et al., 1995) with some of the stacking disorder removed (Murata and Randall, 1975). Temperature and burial depth play an important role in the extent of diagenesis. Therefore, the interpretation of diatomite raises an important second question of whether an unconsolidated sediment can be called diatomite. We suggest here that diatomite, a term widely used for diatom-rich sediments, be used only for consolidated sediments.

We are using a variety of previous studies as a guide, including Murray and Renard (1891), Conger (1942), Bramlette (1946), Murata and Larson (1975), Isaacs (1981c), Pedersen (1981), Kadey (1983), Dean et al. (1985), Mazzullo et al. (1988), Minoura et al. (1996), Inglethorpe (1993), Lemons (1996), Akin et al. (2000), Owen (2002), Moyle (2003), and Schnurrenberger (2003). We adopt the recommendation from Mazzullo et al. (1988), in which any sediment material that is present in the sediment with more than 10% of the composition is considered as a modifier and, therefore, it should be stated in the name. The following definitions of

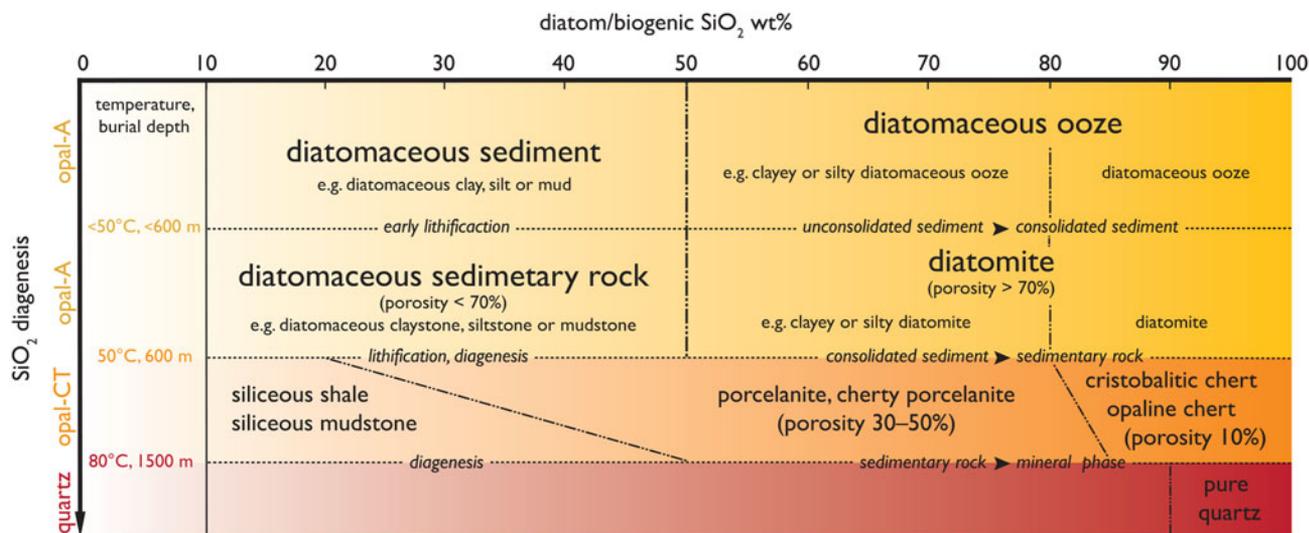


Figure 1. Proposed classification of sediment containing diatoms. The weight percentage (wt%) of diatom/biogenic SiO_2 contained in sediments is on the x-axis and their transformations as a function of burial depth appears on the y-axis.

diatomaceous ooze and diatomite are proposed in Figure 1 and we suggest nomenclature for defining related materials rich in diatom silica.

The principal name describing unconsolidated sediment containing diatom frustules with more than 50% of sediment weight is *diatomaceous ooze*. Oozes have a content of 50 to 80 wt% of diatom SiO_2 together with another major sediment component, such as clay or silt. This major sediment modifier should be stated in the name, e.g., *clayey* or *silty diatomaceous ooze*. The stand-alone principal name *diatomaceous ooze* is used for ooze composed of more than 80 wt% of diatoms without any major sediment modifier. Unconsolidated sediment with diatom frustules less than 50% of sediment weight should be called *diatomaceous sediment* using supporting sediment terminology, such as *diatomaceous clay*, *diatomaceous silt*, or *diatomaceous mud*, to describe the matrix. Sediments with less than 10 wt% of diatom SiO_2 are named based on the clastic sediment classification without reference to diatoms.

Diatomite is defined as a siliceous/opaline consolidated sediment with a diatom SiO_2 content higher than 50% of sediment weight and porosity higher than 70%. As diatomites are often composed of 50–80% by weight of diatom SiO_2 , the major sediment modifier material must be named with the adjectives describing the sediment matrix components, such as clay or silt, for example *clayey* or *silty diatomite*. Further, we suggest that the stand-alone principal name *diatomite* is used strictly for consolidated sediment with more than 80% of sediment weight comprised of diatom SiO_2 . Chrysophyte cysts, radiolarians, and sponge spicules can be a minor part of the sediment (Bramlette, 1946). Consolidated sediments containing less than 50% sediment weight of diatom SiO_2 are then named based on the prevalent sediment component, e.g., *diatomaceous claystone*, *siltstone*, or *mudstone*. The porosity of these consolidated sediments should be lower than 70%. Diatomite and diatomaceous sediments are formed

at low temperatures and pressures (<50°C and burial depth <600m). Significant recrystallization occurs at higher temperature and burial depths.

The boundary at 50% by weight of diatom/biogenic SiO_2 , e.g., between diatomaceous sediment and diatomaceous ooze (unconsolidated sediment) or between diatomaceous sedimentary rock and diatomite (consolidated sediment), as shown in Figure 1, is consistent with the widely used Ocean Drilling Program marine sediment classification system (Mazzullo et al., 1988), as well as the commonly used sedimentological nomenclature practices described by e.g., Schnurrenberger et al. (2003). The boundary at 80% by weight of diatom/biogenic SiO_2 , e.g., between clayey or silty diatomaceous ooze and diatomaceous ooze (unconsolidated sediment) or between clayey or silty diatomite and diatomite (consolidated sediment), is based on the industrial and commercially used diatomite, which has a diatom content higher than 80% by weight.

At burial depths between 600 to 1500 m and temperatures ranging from 50 to 80°C, as found for example in the Monterey Formation, diatomite is transformed into opaline chert with a low porosity of around 10% (Isaacs, 1981b, 1981c). *Chert* is then defined as sedimentary rock with more than 80% of sediment weight originating from biogenic SiO_2 (predominantly diatom SiO_2). Diagenetic processes transform the biogenic SiO_2 (chrysophyte cysts, diatoms, radiolarians, and sponge spicules), and the original source of SiO_2 is difficult to determine, although fossils can be found in cherts (Pessagno and Newport, 1972). For this reason, we use biogenic SiO_2 in our definition of chert instead of the wt% of diatom SiO_2 . Diatomites and diatomaceous sediments containing, for example, clay or silt in combination with between 20 and 85 wt% biogenic SiO_2 , exposed to deeper burial depths and higher temperatures are altered to porcelanite or cherty porcelanite, with a final porosity between 30 and 50%, depending on the clay/silt content (Bramlette, 1946; Isaacs, 1981a; MacKinnon, 1989).

Sediments that contain less than 20 to 50 wt% biogenic SiO₂ under these conditions form shales, siliceous shales or siliceous mudstones, depending on the mineral content (Isaacs, 1981a).

With increasing temperatures and burial depth, the final stage of alteration is reached at depths between 1500 to 2000 m and temperatures above 80°C; these conditions result in recrystallization of opal-CT into quartz (Murata and Larson, 1975; Isaacs, 1980). The boundaries based on biogenic SiO₂ content between cherts, and especially between porcelanites and shales, are not sharp (Fig. 1). This is because the definitions of these siliceous sedimentary rocks take into account not only porosity and content of biogenic SiO₂ but also hardness, fracturing patterns, and other petrological criteria (Isaacs, 1981a), as diagenesis changes the properties of the sedimentary rock. We summarize our suggested classification of diatomaceous sediments and sedimentary rocks based on diatom/biogenic SiO₂ wt% in Figure 1.

In summary, there are inconsistencies in the present use of the term diatomite and diatomaceous sediment throughout the literature. Based on our definition, unconsolidated lake or marine sediments that have not undergone any diagenesis (burial depth and temperature, as defined above), such as those of Holocene age, should not be referred to as diatomite, but instead should be called diatomaceous ooze. We encourage scientists working with diatomaceous sediments and sedimentary rocks to apply the definitions used here to ensure consistent use of terminology and hence comparability among studies. For sediment types that do not fit within the parameters that we have defined (Fig. 1), we suggest that all sediment components (mineralogy, texture, and porosity) be described in detail. We believe that proper naming will bring clarity to future studies focused on diatomaceous sediments and diatomite.

ACKNOWLEDGMENTS

This work was supported by The Royal Physiographic Society in Lund, by the Center for Geosphere Dynamics (UNCE/SCI/006) and by the Charles University project (GA UK 40217) to PZ, the Swedish Research Council to DJC, and NSF EAR-1514814 to SCF.

REFERENCES

- Akin, S., Schembre, J.M., Bhat, S.K., Kovscek, A.R., 2000. Spontaneous imbibition characteristics of diatomite. *Journal of Petroleum Science and Engineering* 25, 149–165.
- Alexander, G.B., Heston, W., Iler, R.K., 1954. The solubility of amorphous silica in water. *The Journal of Physical Chemistry* 58, 453–455.
- Battarbee, R.W., Jones, V.J., Flower, R.J., Cameron, N.G., Bennion, H., Carvalho, L., Juggins, S., 2002. Diatoms. In: Smol, J. P., Birks, H. J. B., Last, W. M., Bradley, R. S., and Alverson, K. (Eds.), *Tracking Environmental Change Using Lake Sediments Volume 3: Terrestrial, Algal, and Siliceous Indicators*. Springer, Dordrecht, pp. 155–202.
- Barker, P.A., Hurrell, E.R., Leng, M.J., Plessen, B., Wolff, C., Conley, D.J., Keppens, E., Milne, I., Cumming, B.F., Laird, K.R., Kendrick, C.P., Wynn, P.M., Verschuren, D., 2013. Carbon cycling within an East African lake revealed by the carbon isotope composition of diatom silica: a 25-ka record from Lake Challa, Mt. Kilimanjaro. *Quaternary Science Reviews* 66, 55–63.
- Bradbury, J.P., Krebs, W.N., 1995. Fossil continental diatoms: paleolimnology, evolution, and biochronology. In: *Siliceous Microfossils* (Short Courses in Paleontology 8). 1995. The Paleontological Society, Knoxville, Tennessee, pp. 119–138.
- Bramlette, M., 1946. The Monterey Formation of California and the Origin of its Siliceous Rocks. Professional Paper 212. U.S. Geological Survey, Washington.
- Clarke, J., 2003. The occurrence and significance of biogenic opal in the regolith. *Earth-Science Reviews* 60, 175–194.
- Conger, P.S., 1942. Accumulation of diatomaceous deposits. *Journal of Sedimentary Research* 12, 55–66.
- Conley, D., Schelske, C., 2001. Biogenic silica. In: Smol, J., Birks, H.J., Last, W., Bradley, R., and Alverson, K. (Eds.), *Tracking Environmental Change Using Lake Sediments, Volume 3: Terrestrial, Algal, and Siliceous Indicators*, Developments in Paleoenvironmental Research. Springer, Dordrecht, pp. 281–293.
- Conley, D.J., 1988. Biogenic silica as an estimate of siliceous microfossil abundance in Great Lakes sediments. *Biogeochemistry* 6, 161–179.
- Dean, W. E., Leinen, M., Stow, D. A. (1985). Classification of deep-sea, fine-grained sediments. *Journal of Sedimentary Petrology* 55(2), 250–256.
- DeMaster, D.J., 1981. The supply and accumulation of silica in the marine environment. *Geochimica et Cosmochimica Acta* 45, 1715–1732.
- Dunham, R.J., 1962. Classification of Carbonate Rocks According to Depositional Texture. In: Ham, W.E. (Ed.), *Classification of Carbonate Rocks*, AAPG, Tulsa, pp. 108–121.
- Flower, R.J., 2013. Diatom Methods! Diatomites: Their Formation, Distribution, and Uses. In: Elias S.A. and Mork, C.J. (Eds.), *Encyclopedia of Quaternary Science*, Elsevier, Amsterdam, pp. 501–506.
- Frings, P.J., Clymans, W., Jeppesen, E., Lauridsen, T.L., Struyf, E., Conley, D.J., 2014. Lack of steady-state in the global biogeochemical Si cycle: emerging evidence from lake Si sequestration. *Biogeochemistry* 117, 255–277.
- Harwood, D.M., Nikolaev, V.A., Winter, D.M., 2007. Cretaceous records of diatom evolution, radiation, and expansion. *The Paleontological Society Papers* 13, 33–59.
- Inglethorpe, S.D.J., 1993. Industrial minerals laboratory manual: Diatomite. Technical report, British Geological Survey, Nottingham.
- Isaacs, C.M., 1980. *Diagenesis in the Monterey Formation examined laterally along the coast near Santa Barbara, California* (Vol. 80, No. 606). Department of Geology, Stanford University.
- Isaacs, C.M., 1981a. Field characterization of rocks in the Monterey Formation along the coast near Santa Barbara, California. In: Isaacs, C.M. (Ed.), *Guide to the Monterey Formation in the California coastal areas, Ventura to San Luis Obispo*. Pacific Section AAPG Spec. Publ., 52, pp. 39–53.
- Isaacs, C.M., 1981b. Outline of diagenesis in the Monterey Formation examined laterally along the Santa Barbara coast, California. In: Isaacs, C.M. (Ed.), *Guide to the Monterey Formation in the California coastal areas, Ventura to San Luis Obispo*. Pacific Section AAPG Spec. Publ., 52, pp. 25–38.
- Isaacs, C.M., 1981c. Porosity reduction during diagenesis of the Monterey Formation, Santa Barbara coastal area, California. In: Garrison, R.E., Douglas R.G., Pisciotto K.E., Isaacs, C.M. and Ingle J.C. (Eds.), *The Monterey Formation and Related Siliceous*

- Rocks of California*. Pacific Section SEPM Spec. Publ., 15, pp. 257–272.
- Kadey, F.L., Jr, 1983. Diatomite. *Industrial Rocks and Minerals* 1, 677–708.
- Lemons, J.F.J., 1996. Diatomite. Minerals Yearbook - Metals and Minerals. United States Geological Survey, Reston, available online: <https://www.usgs.gov/centers/nmic/diatomite-statistics-and-information>.
- MacKinnon, T.C., 1989. Origin of the Miocene Monterey Formation in California. In: MacKinnon, T.C., Randall, J.W. and Garrison, R.E. (Eds.), *Coal and Hydrocarbon Resources of North America*; Volume 1, Oil in the California Monterey Formation, Field Trips for the 28th International Geological Congress, Washington, DC, Am. Geophys. Union, T311, pp. 1–10.
- Mazzullo, J.M., Meyer, A., Kidd, R.B., 1988. New sediment classification scheme for the Ocean Drilling Program. In: Mazzullo, J.M., Graham, A.G. (Eds.), *Handbook for shipboard sedimentologists*. Texas A&M University, ODP, Technical Note, 8, pp. 45–67.
- Minoura, K., Susaki, T., Horiuchi, K., 1996. Lithification of biogenic siliceous sediments: Evidence from Neogene diatomaceous sequences of northeast Japan. *Sedimentary Geology* 107, 45–59.
- Mortlock, R.A., Froelich, P.N., 1989. A simple method for the rapid determination of biogenic opal in pelagic marine sediments. Deep Sea Research Part A. *Oceanographic Research Papers* 36, 1415–1426.
- Moyle R.P., Dolley P.T., 2003. Chapter D: With or without Salt—A comparison of marine and continental-lacustrine diatomite deposits. In: Bliss J.D., Moyle, P.R., Long, K.R. (Eds.), *Contributions to Industrial-Minerals Reserch*. Bulletin 2209–D. US Geological Survey. <https://doi.org/10.2172/820335>
- Murata, K.J, Larson, R.R., 1975. Diagenesis of Miocene siliceous shales, Temblor Range, California. *Journal of Research of the U.S. Geological Survey* 3, 553–566.
- Murata, K. J., & Randall, R. G. (1975). Silica mineralogy and structure of the Monterey shale, Temblor Range, California. *Journal of Research of the U.S. Geological Survey*, 3, 567–572.
- Murray, J., Renard, A.F., 1891. *Report on Deep-sea Deposits based on the Specimens collected during the Voyage of H.M.S. Challenger in the years 1872 to 1876*. HM Stationery Office, Edinburgh.
- Opfergelt, S., Eiriksdottir, E.S., Burton, K.W., Einarsson, A., Siebert, C., Gislason, S.R., Halliday, A.N., 2011. Quantifying the impact of freshwater diatom productivity on silicon isotopes and silicon fluxes: Lake Myvatn, Iceland. *Earth and Planetary Science Letters* 305, 73–82.
- Owen, R.B., 2002. Sedimentological characteristics and origins of diatomaceous deposits in the East African Rift System. *SEPM Special Publication* 73, 233–246.
- Pedersen, G.K., 1981. Anoxic events during sedimentation of a Palaeogene diatomite in Denmark. *Sedimentology* 28, 487–504.
- Pessagno, E.A., Jr., Newport, R.L., 1972. A technique for extracting radiolaria from radiolarian cherts. *Micropaleontology* 18, 231–234.
- Pettijohn, F., 1949. *Sedimentary Rocks*. 2nd ed. Harper and Row, New York.
- Plunkett, G.M., Whitehouse, N.J., Hall, V.A., Brown, D.M., Baillie, M.G.L., 2004. A precisely-dated lake-level rise marked by diatomite formation in northeastern Ireland. *Journal of Quaternary Science* 19, 3–7.
- Reynolds B.C., Aggarwal J., André L., Baxter D., Beucher C., Brzezinski M.A., Engström E., et al., 2007. An inter-laboratory comparison of Si isotope reference materials. *Journal of Analytical Atomic Spectrometry* 22, 561–568
- Rice, S.B., Freund, H., Huang, W.L., Clouse, J.A., Isaacs, C.M., 1995. Application of Fourier transform infrared spectroscopy to silica diagenesis: The opal-A to opal-CT transformation. *Journal of Sedimentary Research* 65, 639–647.
- Schnurrenberger, D., Russell, J., Kelts, K., 2003. Classification of lacustrine sediments based on sedimentary components. *Journal of Paleolimnology* 29, 141–154.
- Terzaghi, K., Peck, R.B., Mesri, G., 1996. *Soil Mechanics in Engineering Practice*. 3rd ed. John Wiley and Sons, New York.
- Theriot, E.C., Fritz, S.C., Whitlock, C., Conley, D.J., 2006. Late Quaternary rapid morphological evolution of an endemic diatom in Yellowstone Lake, Wyoming. *Paleobiology* 32, 38–54.
- Wallace, A. R. (2003). Regional geologic setting of Late Cenozoic lacustrine diatomite deposits, Great Basin and surrounding region: overview and plans for investigation. US Department of the Interior, US Geological Survey.
- Wentworth, C.K., 1922. A scale of grade and class terms for clastic sediments. *The Journal of Geology* 30, 377–392.