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ment. In the case of metal-polymer core-shell structures, the fluorescence of the organic dye may be quenched because of complex interactions with the metal core. To study the effect of separation distance on fluorescence quenching, G. Schneider and G. Decher of Institut Charles Sadron and Université Louis Pasteur, along with N. Nerambourg, R. Praho, M. Werts, and M. Blanchard-Desce of Université de Rennes, have described an approach in which up to 20 polyelectrolyte layers (the last layer being fluorescently labeled) were coated on gold nanoparticles. The core-shell distance was altered by tailoring the thickness of adsorbed, nonfluorescent polymers, which were used as spacer layers in the core-shell structure.

As reported in the March 8 issue of *Nano Letters* (10.1021/ nl052441s; p. 530), gold particles (diameter, 13 nm) were synthesized by the reduction of tetrachloroauric acid with trisodium citrate. Spacer layers of polyallylamine hydrochloride (PAH) and polystyrene sulfonate (PSS) were adsorbed consecutively using electrostatic layer-by-layer assembly to obtain gold nanoparticles coated with 2, 10, and 20 layers of Au(PAH/PSS)_n. Fluorescein

isothiocyanate (FITC) and lissamine rhodamine B sulfonyl chloride (LISS) were covalently attached to PAH to obtain PAH-FITC and PAH-LISS, which were then adsorbed onto the Au(PAH/PSS)_n core–shell particles, followed by the adsorption of a final PSS layer. After each coating step, the particles were washed thoroughly by successive centrifugation/re-suspension cycles. Transmission electron microscopy and *in situ* UV-visible absorption spectra were used to characterize the structural and photophysical properties of the nanoparticle dispersions.

The researchers said that the addition of dye-functionalized polyelectrolytes does not perturb layer growth or the aggregation state of the solution. For n > 5 layers, the surface plasmon resonance from the gold particles reaches a fixed position and intensity, 534 nm in the case of PAH-FITC and 533 nm for PAH-LISS, and obscures the dye-related absorption bands. The surface-plasmon band was removed by slow dissolution of the gold core by adding cyanide ions to the solution. Dissolution of the gold core is accompanied by a strong increase in fluorescence, and reveals the weak dye-related absorption band.

Fluorescence quenching is distance-dependent and remained significant 8 nm from the metal core. The fluorescence quenching seems to be caused by a decrease in radiative transition probabilities induced by the gold core. The researchers said that despite quenching, dye fluorescence remains sufficiently bright for diagnostic and sensing applications.

JEREMIAH T. ABIADE

Highly Porous Inorganic Thin Films Formed Using Polymer Microsphere Templates

The use of colloidal templates as self-assembled building blocks for fabricating ordered submicrometer structures of various materials has been explored in recent years. These templates have been used to fabricate photonic crystals and synthesize catalysts and sensors. However, synthesis of compounds with complex compositions using this method has been difficult due to the wet chemical techniques used and the ability of the chemicals to infiltrate the voids between the colloids with the precursors. A study done at the Massachusetts Institute of Technology has now demonstrated the successful application of



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microsphere templating for fabricating macroporous CaCu₃Ti₄O₁₂ (CCTO) thin films using pulsed laser deposition (PLD) on poly(methyl methacrylate) (PMMA) templated substrates. The polymer microspheres are removed by thermal decomposition, and the residual inorganic films have unique morphologies comprising hollow hemispheres of the inorganic material with large surface areas and surface-tovolume ratios. I.-D. Kim, A. Rothschild, and co-workers reported the preparation of the CCTO thin films, as well as gas-sensitivity measurements for possible sensor applications, in the February issue of Nano Letters (DOI: 10.1021/nl051965p; p. 193).

Given its extraordinarily high dielectric constant and nonlinear I-V characteristics, CCTO was expected to make an excellent sensor when interacting with reactive gases. To prepare the CCTO films, an aqueous solution of 800-nm diameter PMMA microspheres was dripped onto Al₂O₃ substrates which had interdigitated Pt electrode arrays and then dried overnight. CCTO films were then deposited onto the microsphere-templated substrate using PLD. The samples were calcined in air at 800°C for several hours to remove the PMMA microspheres. The resultant intact films were 80-90 nm thick and consisted of a monolayer of hollow CCTO hemispheres with ~800-nm diameters, as shown in Figure 1. Good chemical bonding between the film and substrate was observed. X-ray diffraction measurements showed that the films were singlephase polycrystalline CCTO.

A network of voids between the film and substrate allowed easy gas access to the internal surfaces of the hemispheres. The specific surface area was twice as large and the surface area four times as large as films on untreated substrates due to the hemispherical morphology and the voids-very favorable for gas sensor applications. Electrical property measurements demonstrated highly nonlinear I-V characteristics with strong thermal activation. The macroporous CCTO films were tested for sensitivity to H₂, CO, and CH₄ gases between 200-400°C and were found to be very sensitive to H₂. In contrast, the sensitivity of films deposited on untreated substrates to H₂ was negligible. The large surface area clearly promoted interactions with the gas while the relatively smaller interfacial area reduced the adverse effects of the interface between the film and the substrate. The limit of detection was determined to be about 55 ppm at a temperature of 300°C, comparable to the detection limit of commonly used commercial Taguchi H₂ sensors. The responsivity of

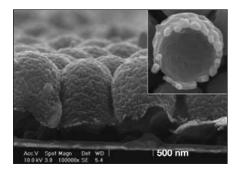


Figure 1. Scanning electron micrograph shows a cross-sectional view of the macroporous CaCu₃Ti₄O₁₂ (CCTO) film formed following calcination at 800°C. The inset shows one hollow CCTO hemisphere. (Courtesy of Avner Rothschild, MIT.)

the macroporous films to H_2 was also found to be reversible.

The researchers said that in view of the simplicity of this approach, which can be applied to a wide range of materials and deposition techniques, and the remarkable improved performance, this processing strategy holds promise for chemical sensors, fuel cells, catalysts, and other electrochemical devices for energy conversion, environmental protection, and other applications. The technique has the ability to control parameters such as porosity, composition, and film thickness as well as form multilayered structures of compounds with complex chemistries and therefore is very versatile.

The research team's future plans include extending this work to other materials commonly used for gas sensing, such as SnO₂, TiO₂, and ZnO, and examining the role of processing conditions on microsphere morphology and correspondingly on the sensing properties. The researchers will also investigate the negligible responsivity of the macroporous films to gaseous CO and CH₄.

GOPAL RAO

CdSe/CdS QD-LEDs Fabricated Using a Thermally Polymerized Hole-Transport Layer

By spin-coating CdSe/CdS quantum dot solutions onto thermally cross-linkable hole-transport layers (HTLs), a team of researchers at the University of Washington fabricated multilayer nanocrystal quantum dot light-emitting diodes (QD-LEDs) with external quantum efficiencies (EQEs) of ~0.8% that exhibit spectrally pure QD emission. This method enables researchers to incorporate high-quality QD layers of varied thickness into devices

with multiple solution-processed organic hole-transport layers and no organic background emission.

In the March issue of Nano Letters (DOI: 10.1021/nl052417e; p. 463), the research team, led by A.K.-Y. Jen and D.S. Ginger, reported progress in the fabrication of efficient CdSe/CdS QD-LEDs. The researchers first spin-coated the cross-linkable HTL, polystyrene-N,N'-diphenyl-N,N'bis(4-n-butylphenyl)-(1,1'-biphenyl)-4,4'diamine-perfluorocyclobutane (PS-TPD-PFCB), onto a plasma-cleaned indium tin oxide (ITO) slide and then heated it in nitrogen to activate the cross-linking process in the HTL. After cross-linking the PS-TPD-PFCB, the researchers spin-coated a layer of CdSe/CdS QDs (average diameter, 4.6 nm) from a chloroform solution onto the HTL. They then deposited the electron transport layer, 1,3,5-tris(Nphenylbenzimidazol-2-yl)benzene (TPBI), on the QD layer by thermal evaporation. They completed the devices by thermal evaporation of the back electrode (Ca/Ag).

The thermally cross-linked HTLs possess a high glass-transition temperature ($T_{\rm g}$ ~ 150°C) and are solvent-resistant, factors which make it easy to add additional device layers such as emissive QDs by solution processing. Using atomic force microscopy (AFM), the researchers found that the surface of the polymerized PS-TPD-PFCB HTL is very smooth and provides a compatible hydrophobic surface for the formation of uniform and pinholefree layers of QDs over comparatively large areas. The researchers compared the QD-LEDs with different CdSe/CdS QD coverage in the active region. They observed that the device efficiency increases with increasing QD coverage until one monolayer (ML) of dots has been deposited; then it decreases rapidly to 0.1% when the QD layer is more than 3 ML thick. The researchers also observed a substantial decrease in EQE with increasing current density. They attributed these phenomena to slow dot-dot charge transport and a carrier imbalance or charge-induced quenching, respectively. The researchers found they could further improve the EQE of the LEDs by using a second (cross-linkable) HTL on top of the PS-TPD-PFCB layer. The EQE of these QD-LEDs is improved to 0.8% at a brightness of 100 cd/m^2 and the maximum brightness is improved to more than 1000 cd/m². The researchers suggested that optimizing the energy levels and better balancing carrier transport and injection into the QD emissive layer may further improve the performance of these QD-LEDs.

Tao Xu