Systematic Transmission Electron Microscopy Study Investigating Lithium and Magnesium Intercalation in Vanadium Oxide Polymorphs

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Magnesium-ion based batteries promise a competitive alternative to conventional lithium-ion battery technology. Batteries combining Mg metal anode with a suitable intercalation-based cathode can offer much higher volumetric energy density, as well as significant cost and safety benefits over lithium ion batteries. Recent first-principles and experimental reports have established that orthorhombic α -V₂O₅ is a promising intercalation cathode for Mg ion batteries. However, several crucial aspects of the intercalation phenomenon, such as the specific intercalation sites for Mg within α -V₂O₅ or the formation of different phases upon Mg insertion into α -V₂O₅ remain unclear. Further systematic characterization of the Mg intercalation behaviour is therefore required.

This contribution will focus on systematic investigation of Mg intercalation into α -V₂O₅ by combining aberration-corrected scanning transmission electron microscopy (STEM) imaging, electron diffraction, electron energy loss (EEL) and energy dispersive x-ray spectroscopy (XEDS). More specifically, we will present a comparison of Mg insertion sites in two different samples: i) electrochemically cycled α -V₂O₅ cathode in a prospective full cell vs Mg metal anode and ii) chemically synthesized MgV₂O₅ sample. In the case of electrochemically cycled α -V₂O₅, our results determine the Mg intercalation sites and it is concluded that this sample exhibits the local formation of the ϵ -Mg_{0.5}V₂O₅ phase, as predicted by earlier first-principles density functional theory (DFT) calculations [1]. Figure 1a) and b) present atomic resolution high-angle annular dark-field (HAADF) and annular bright-field (ABF) images, respectively, for the electrochemically-cycled orthorhombic α -V₂O₅ cathode. Simulated HAADF and ABF images for the DFT predicted ϵ -Mg_{0.5}V₂O₅ phase are overlaid on the experimental STEM images. The structural model for the ϵ -Mg_{0.5}V₂O₅ phase is shown in Fig 1(c). We will also show that the chemically synthesized sample presents the δ -MgV₂O₅ phase [2].

Recent theoretical calculations have also predicted that the migration barrier for ionic intercalation can be decreased by exploiting different anion coordination environments in metastable vanadium oxide polymorphs, such as ζ -V₂O₅ [3]. Figure 2a) presents atomic-resolution HAADF image for ζ -V₂O₅ nanowires clearly showing the heavier V atoms and elucidating the tunnel structure for this novel polymorph; a structural model for the ζ -V₂O₅ phase is presented in Figure 2b). We have investigated the lithium intercalation in this tunnel-structured ζ -V₂O₅ polymorph, and will focus on showing that ζ -V₂O₅ nanowires show much better Li-cycling properties (i.e. reversibility) compared to orthorhombic α -V₂O₅ [4]. Moreover, Mg intercalation into ζ -V₂O₅ nanowires will be investigated in detail, comparing electrochemical performance at both low and high temperature cycling followed by systematic STEM characterization. The results obtained for this novel polymorph ζ -V₂O₅ polymorph, such as ϵ -V₂O₅ will also be tested for their ability to intercalate Li or Mg [5].

References:

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[3] A. Parija et al, Chemistry of Materials 28 (2016), p. 5611.

[4] A. Mukherjee *et al*, Direct observation of reversible topotactic lithium intercalation in tunnel structure ζ -V₂O₅ nanowire cathode, (*in preparation*)

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Figure 1. (a) Filtered atomic-resolution HAADF image of electrochemically cycled α -V₂O₅ cathode, simulated image overlaid (b) Filtered ABF image, simulated image overlaid (c) Structural model for DFT predicted ϵ -Mg_{0.5}V₂O₅ phase in [001] zone





Figure 2. (a) Filtered atomic-resolution HAADF image for pristine ζ -V₂O₅ nanowire (b) Structural model for ζ -V₂O₅ in $\overline{[150]}$ orientation

<u>1 nm</u> tered atomic-resolution HAADF image image overlaid (b) Filtered ABF image, edicted ε-Mg_{0.5}V₂O₅ phase in [001] zone