Understanding Mineral Carbonate Formation under Supercritical Conditions using Electron Microscopy and Atom Probe Tomography

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Olivines, a significant constituent of basaltic rocks, have the potential to immobilize permanently CO₂ after it is injected in the deep subsurface, due to carbonation reactions occurring between CO₂ and the host rock. To investigate the reactions of olivine minerals, forsterite (Mg₂SiO₄) and fayalite (Fe₂SiO₄), with supercritical CO₂ (scCO₂) and formation of mineral carbonates, experiments were conducted at temperatures of 35 °C to 80 °C, 90 atm pressure and anoxic conditions for favalite study. For every temperature, the dissolution of forsterite and fayalite was examined both in the presence of liquid water and H₂O-saturated scCO₂. It has been recognized that interactions with near to water-saturated non-aqueous fluids are of prime importance in understanding mineralization reactions since the introduced CO₂ is likely to contain water initially or soon after injection and the supercritical CO₂ (scCO₂) is less dense than the aqueous phase which can result in a buoyant scCO₂ plume contacting the isolating caprock. The experiments were conducted in a high-pressure batch reactor at reaction time extending up to 85 days. The newly formed products were characterized ex-situ using a comprehensive suite of bulk and surface characterization techniques: X-ray diffraction, Scanning Electron Microscopy coupled with Focused Ion Beam, and High Resolution Transmission Electron Microscopy and Atom Probe Spectroscopy. Furthermore nucleation of Fe, Mg-carbonates during reaction of orthosilicates with CO₂ were studied via in situ TEM using a fluid cell and holder.

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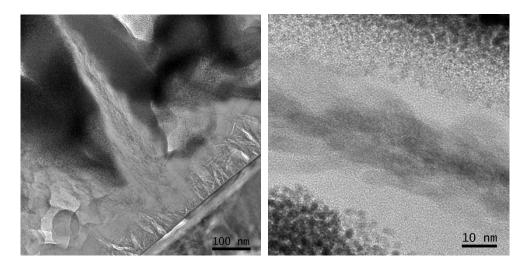


Figure 1. (a) cross section of platelet particles displaying a core identified as hematite; (b) high resolution image of the hematite particle obtained on [441] zone axis.

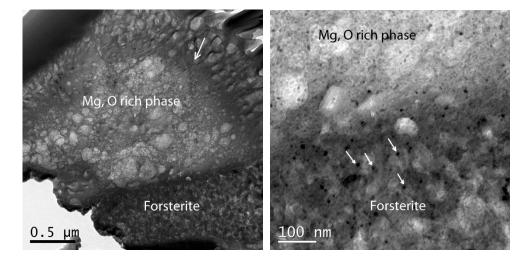


Figure 2. (a) Detail depiction of the interface between the nanocrystalline forsterite and the new precipitate phase. Intergrowth of the new precipitate phase within the forsterite is emphasized with the arrows (b) Atomic resolution view of the new precipitate phase. High density of planar faults was identified in many grains of the new precipitate phase.