## **Connecting Phase Stability to the Grain Growth Behavior of Ni-W Alloys**

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Nanocrystalline materials are beneficial due to their promising mechanical properties. However, selecting proper alloy compositions to limit unwanted grain growth has become a challenge because of increasing difficulties in understanding nanocrystalline thermodynamics and kinetics. There is a strong dependence on accurate phase diagrams to correctly predict at which compositions certain thermal stability mechanisms are most likely active. Regardless of the stability mechanism, proper knowledge of the phase diagram is paramount to choose the best alloy composition.

Ni-W alloys are classified as thermodynamically stable materials because W-segregation lowers the specific energy of the grain boundaries and reduces the driving force for grain growth [1]. However, the binary Ni-W phase diagram has been questioned in regard to the existence of high W-content intermetallics, namely NiW and NiW<sub>2</sub> [2,3]. If the phase diagram is indeed inaccurate, some phases may form before they are anticipated and de-stabilize the microstructure. Considering that the phase diagram is an essential predictive tool for any nanocrystalline material, this work has taken a comprehensive approach to characterize several heat-treated Ni-W alloys to aide in proper design of this nanocrystalline system.

Electrodeposited and DC magnetron sputtered alloys of differing W content were annealed in a reducing  $N_2$ -5% $H_2$  atmosphere at several temperatures to track Ni-W phase evolution. Aberration-corrected scanning transmission electron microscopy (STEM) was the primary tool to characterize the microstructures and identify inconsistences of the binary phase diagram.

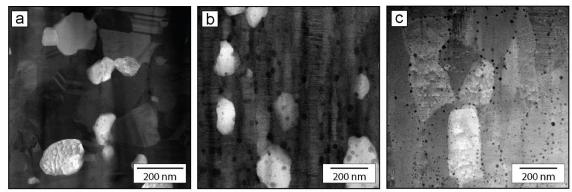
First and foremost, several unexpected impurity phases that are not found on the phase diagram were identified. In agreement with Cury et al (2009), the intermetallic phase NiW was not observed in alloys exceeding 20 at%W, but instead, the impurity carbide phase Ni<sub>6</sub>W<sub>6</sub>C was found [3]. Furthermore, the Ni<sub>6</sub>W<sub>6</sub>C was identified in both the electrodeposited and sputtered alloys annealed at 700 °C, which suggests that the Ni<sub>6</sub>W<sub>6</sub>C precipitation may be more widespread than currently realized. Figures 1a (electrodeposited) and 1b (sputtered) highlight Ni<sub>6</sub>W<sub>6</sub>C via bright contrast in STEM-HAADF images. Interestingly, the electrodeposited STEM-HAADF image featured a pure W phase as well. To prove that the impurity phase was Ni<sub>6</sub>W<sub>6</sub>C and not NiW, STEM-EDS, atomic resolution imaging and selected area diffraction patterns confirmed that the composition and crystal structure was indeed Ni<sub>6</sub>W<sub>6</sub>C [4]. Besides Ni<sub>6</sub>W<sub>6</sub>C, W-rich oxide particles were consistently observed in electroplated and sputtered alloys. In electrodeposited alloys, oxide particles were measured to be of appropriate size and volume fraction to pin grain boundaries and limit grain growth [5]. In sputtered alloys, the oxide particles were observed on the columnar grain boundaries, as seen in Figure 1b. Perhaps the most important observation regarding the nanoscale oxide particles was that W-segregation to oxide phase boundaries was consistently observed. Thus, an indirect thermal stability mechanism of phase boundary segregation can reduce grain growth rates because the particles are now resistant to coarsening. Lastly, and specific to the sputtered alloys, Ar pores decorated the columnar grain boundaries. Similar to the oxide particles, it is probable that the Ar pores act as pinning agents to slow grain growth.

In contrast to NiW, the intermetallic Ni<sub>4</sub>W was confirmed to exist. In addition, the disordering transformation that occurs between 700 °C and 1000 °C was confirmed. Figure 2 shows that at 700 °C there are two Ni<sub>4</sub>W variants that are clearly observable, and after heating above 1000 °C, the W atoms are dissolved back into a disordered solid solution. Older reports have also observed Ni<sub>4</sub>W chemical ordering, but they did not have the capability to characterize the atomic structure [6]. Interestingly, alloys within the Ni(W) and Ni<sub>4</sub>W two-phase region exhibited abnormal grain growth at 700 °C.

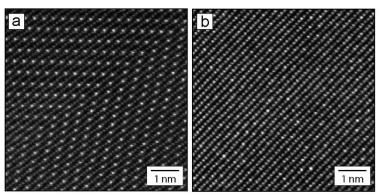
Overall, this study has re-evaluated the Ni-W binary phase diagram, identified the most likely impurity phases, and connected the new findings to nanocrystalline grain growth behavior.

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**Figure 1.** STEM-HAADF images that aided in the identification of impurity phases in electrodeposited and sputtered Ni-W alloys: (a)  $Ni_6W_6C$ , (b)  $Ni_6W_6C$  and WOx, and (c) Ar pores.



**Figure 2.** STEM-HAADF images that show the ordering transition of Ni<sub>4</sub>W after heat treating in N<sub>2</sub>-5%  $H_2$  for 24 hrs.