

## Preparation of a Model Platinum/Gamma-Alumina Catalyst for *in situ* Environmental TEM Experiments

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Platinum/gamma-alumina (Pt/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub>) is an important catalyst system used in a variety of industrial applications, such as fuel cells, petroleum reformers, and catalytic converters [1]. Correlating theoretical and experimental studies of Pt/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub> has been hindered by the non-uniform crystallinity of commercially available  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> [2]. Synthesizing a well-defined, model system of Pt/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub> is crucial for bridging the gap between experiment and theory by enabling direct comparison of experimental results and theoretical simulations [3]. In this work, we present our synthesis of a model catalyst consisting of a single-crystal, defect free  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> thin film decorated with well-defined Pt nanoparticles evenly dispersed on the  $\gamma$ -Al<sub>2</sub>O<sub>3</sub>. Cross-sectional TEM samples made from the model system will be employed for *in situ* environmental transmission electron microscopy (ETEM) experiments, since observing Pt/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub> during reaction under relevant gas and thermal conditions is crucial to understanding the important structural features impacting the catalytic activity of the Pt nanoparticles.

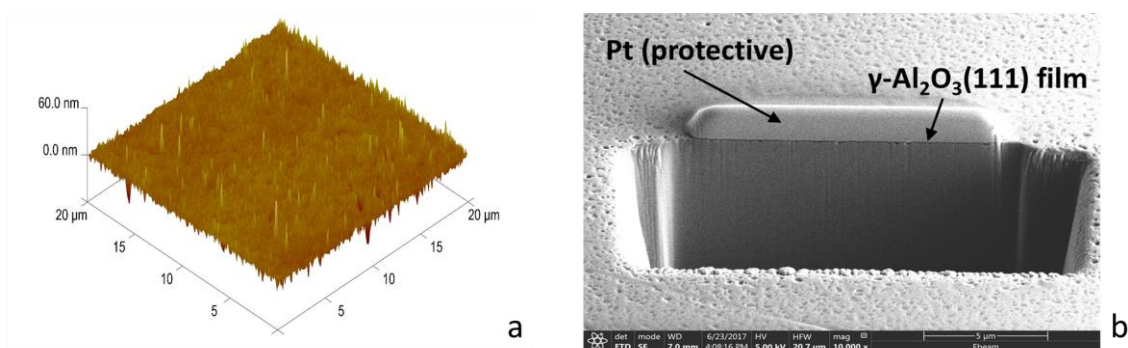
Defect-free, single-crystal  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> (111) was synthesized through controlled oxidation of single-crystal NiAl (110) [2], the orientation of which was verified by X-ray diffraction (XRD). The NiAl surface was then polished through mechanical grinding techniques to promote homogenous growth of defect-free  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> (111) with root mean squared (RMS) surface roughness of 5.21 nm, as measured using atomic force microscopy (AFM). The NiAl was then oxidized in a tube furnace at 750°C for two hours in dry air. The resulting single-crystal  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> was determined to be of the (111) orientation via XRD, and the surface roughness was measured by AFM to be less than 6 nm RMS (Figure 1a).

A dual beam, focused ion beam-scanning electron microscope (FIB-SEM) was used to prepare a cross-section from the sample, exposing the NiAl/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub> interface (Figure 1a). From the SEM image of the cross-section, the  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> layer was determined to be about 60nm thick. 1-1.5 nm diameter Pt nanoparticles were then deposited onto the  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> surface by electron beam evaporation. A cross-sectional TEM sample was made from the Pt/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub>/NiAl (Figure 2). Deposition of monodisperse, shape-controlled, 1 nm Pt nanoparticles prepared by the inverse micelle technique [4] onto the model  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> surface is underway.

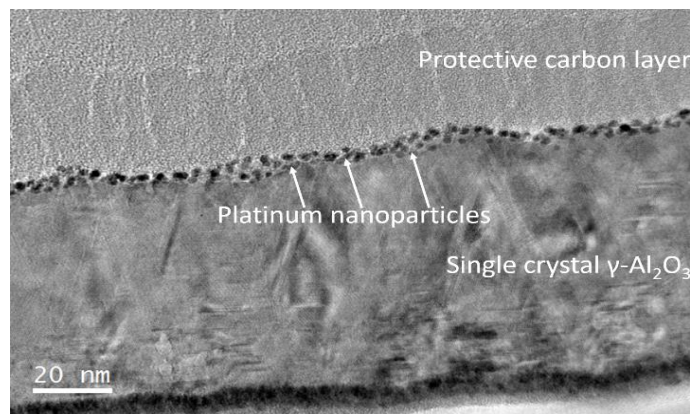
In addition to the model catalyst, Pt/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub> sample were prepared using commercially available polycrystalline  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> for comparison. Polycrystalline  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> powder was solution drop-cast onto a copper TEM grid, upon which 1-1.5 nm Pt nanoparticles were subsequently deposited via electron beam evaporation. *In situ* environmental TEM experiments are being performed on both the model and non-uniform Pt/ $\gamma$ -Al<sub>2</sub>O<sub>3</sub> catalysts to understand the role of the  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> support on Pt nanoparticle behavior under relevant reaction gas and heating conditions. [5]

## References:

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 [3] M. Willinger, *et al.*, *Angewandte Chemie International Edition* **23** (2014), pp. 5998–6001.  
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**Figure 1:** (a) AFM surface roughness measurement of one of the  $\gamma\text{-Al}_2\text{O}_3$  films. (b) Cross-sectional SEM image of the NiAl/ $\gamma\text{-Al}_2\text{O}_3$  interface, lifted from the sample in (b).



**Figure 2:** TEM micrograph the Pt/ $\gamma\text{-Al}_2\text{O}_3$  interface the resulting lift-out, showing the dispersion of nanoparticles across the surface.