Extraction and Characterization of chemical constituents present in Cuphea aequipetala and their properties.

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INTRODUCTION

Mexico has a great importance in terms of medicinal plants due to its cultural and biological diversity while the studies of medicinal plants oriented towards quality control are scarce. One of the plants of American origin used in herbalism from Mexico is Cuphea aequipetala (C. aequipetala), currently this plant is on the list of species of ethnobotanical use in Mexico from the Herbal Extrapharmacopoeia. Taxonomically C.aequipetala is a perennial plant of herbaceous or subfruitful habit. It is a plant of wide distribution in Mexico, mainly in the center and south areas (popularly known as "hierba de cancer" but Márquez et al. (1999) y Martínez (1979) mentioned the other common name of the same as; alcáncer, apanchola, atlanchane, pegamosca, hierba de la calavera, mirto, trompetilla, ajam bomol, petalillo, hierba del ángel, guinda, cachancillo, alfilerillo, miel de abeja, motoye, pantillo, florecita roja, petalito, tatalito, siltic y zapotillo.

C.aequipetala has been used for the treatment of cancer, inflammation, pain, diarrhoea, fever, wounds and stomach infections. Today, many countries use plants to treat different maladies, such is the case of aqueos and organic extracts of C. aequipetala have shown gastroprotective actions, cytotoxic activity against cancer cell lines, anti-lipase activity and antioxidant effects[4-7].

The investigation of medicinal plants using frontier technologies is now being reconsidered to be a feasible approach for discovering novel bioactive agents to solve widespread public health problems [7-9]. In this way increasing interest has been focused into the procedures for extract preparation and chemical constitutes characterizations because of their influence in the yield and modification of the activity of compounds. As a first approach in this study, the extracts obtained from C. aequipetala were analyzed by different methods. Being the MS-DART, FTIR, and Raman spectroscopy the techniques used for the chemical characterization of these.

MATERIALS AND METHODS

2.1 Plant material

The samples of C. aequipetala was obtained dry from a local vendor in La Piedad, Michoacan, who obtained it directly from indigenous people in the Purepecha area. The dried sample was placed in the

oven at 45°C for 48 hours to remove the remaining moisture in the material. Subsequently, the humidity percentage was obtained, followed by this the material was subjected to a micromilling procedure in capsules with zirconium coating, to avoid contamination in the simples, these capsules were placed in a micromill equipment MM 400 Retsch, to obtain a particle size of 63µm, this to homogenize and increase the effectiveness in the extraction process.

2.2 Extraction methods

The extracts of C. aequipetala were obtained by Soxhlet extraction method in four different solvents (ethanol, hexane, high-purity methanol and low purity methanol) for 4 h. The solutions obtained in each case were concentrated by rotary evaporation and their at respective temperature and obtained extracts were collected and conserved for further analyses.

2.3 Analytical techniques

a) Mass spectrometry DART-MS

Direct analysis in real time mass spectroscopy (JEOL-AccuTOF JMS-T100LC (Jeol, Tokyo, Japan)) was applied to study of the present compounds in extracted samples where, Helium was used as the DART source gas for all analyses and polyethylene glicol 600 (PEG) was used for mass calibration.

b) Fourier transform infrarred spectroscopy coupled with attenuated total reflection (FTIR-ATR)

The Fourier transform infrared spectroscopy (FTIR, 110 Frontier MIR/NIR Perkin Elmer) was used to study the functional groups present in extracted samples between 4000-400 cm-1 at resolution 4cm-1, 20 scans were averaged in order to obtain a clear spectra of each sample.

RESULTS

The measurement of molecular compounds obtained by AccuTOFTM-DART® (JEOL-AccuTOF JMS-T100LC (Jeol, Tokyo, Japan) where the obtained compounds are mentioned in Figure 1.

The results shows that the possible extraction of compounds are only in selective solvents, like glucose was only possible in methanol based extraction both in high-pure as well low-pure, which is a very efficient source of energy in human body for daily uses as well on many specific diseases treatments as an energy source, while the ergosterol and epifriedelanol were found in hexane, while we could not be able to find any compound in ethanol based extraction.

The extract obtained using hexane showed two bands of high interests, one with a molecular weight of 396.65 g/mol in the first band would be related to Ergosterol, a sterol and biological precursor of vitamin D, also contains the antitumor properties. Ergosterol peroxide is used to treat breast cancer. The second band was given by a molecular weight of 428.7 g/mol related to Epifriedelanol, which has anticancer properties.

The FTIR results (shown in Figure 2) from the different extracts obtained with ethanol, also pure and recirculated methanol in their corresponding spectra, showed a characteristic band ~3400-3500 cm-1 related to stretching vibration of OH groups. It should be noted that this band is not clearly seen in the extract obtained with hexanol, which can be attributed to the drying and pretreatment times of the sample.

In all spectra, a band is shown around ~1709-1705 cm-1, probably related to: stretching vibration of carboxyl groups, stretching of C = O of flavonoids and lipids. The bands in the range ~1456-1452 cm-1, could be related to CH3, CH2, flavonoids and aromatic rings, where the vibrations would be the bending (δ) vibration of C–H and the stretching vibration of aromatics.

Compounds Obtained		HC-SMP (ultra- pure methanol)	HC-SMR (low- purity methanol)	HC-SME (high- purity ethanol)	HC-SH (high purity hexane)
Glucose C ₆ H ₁₂ O ₆ 180.156 g/mol	HO CH ₂ OH HO OH Glucose	X	X		
Ergosterol C ₂₈ H ₄₄ O 396.65 g/mol	HO H H				Х
Epifriedelanol C ₃₀ H ₅₂ O 428.7 g/mol	HO H				X

Figure 1. Compounds obtained in extracted solvents by DART-MS.

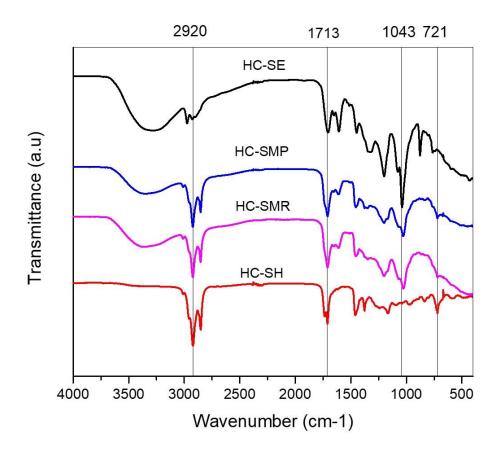


Figure 2. FTIR bands related to the C. aequipetala extract obtained By ethanol (HC-SE), hexane (HC-SH), high-pure methanol (HC-SMP), and low-purity methanol (HC-SMR) and their comparison.

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