

Synthesis and Characterization of ZnO/ α -Fe₂O₃ Nano-Architectures

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One dimensional nanocomposite materials and complex nano-architectures possess unique physical and chemical properties and may find potential applications in nanoscale electronic devices, photovoltaics, sensing, and nanocatalysts. Alpha-Fe₂O₃ (hematite) nanowires are of interest because of their potential applications in magnetic recording and as catalyst for the dehydrogenation of ethylbenzene to styrene [1]. Moreover, α -Fe₂O₃ composite materials have attracted interest as gas sensors [2]. In this paper, we report the recent successful synthesis of complex ZnO/ α -Fe₂O₃ nano-architectures.

The α -Fe₂O₃ nanowires and nanobelts were synthesized by direct thermal oxidation of Fe substrates in a tube furnace [3]. The as-synthesized α -Fe₂O₃ nanowires and nanobelts were then loaded into a tube furnace as the substrate materials. The ZnO source material, located at the center of the tube furnace, was then heated to 1000°C with a flowing gas (mixture of argon and oxygen) which carried the Zn/ZnO molecules to lower temperature regions where they deposited onto the α -Fe₂O₃ nanowires/nanobelts. The substrates occupied a furnace region where the temperature ranged from about 610°C to about 330°C. A field emission SEM equipped with an energy dispersive X-ray spectrometer and a Robinson backscattered electron detector was used to characterize the morphology, composition and size distribution of both the uncoated α -Fe₂O₃ nanowires/nanobelts and the complex ZnO-Fe₂O₃ nano-architectures.

The as synthesized α -Fe₂O₃ nanowires were aligned and grew perpendicular to the substrate with an average length and width of 10.0 μ m and 0.3 μ m, respectively. After deposition of ZnO, the α -Fe₂O₃ nanowires maintained their original growth directions and were coated with thin nanowires growing almost perpendicular to the α -Fe₂O₃ nanowires (Fig. 1); the as synthesized materials were collected in the temperature region from about 330°C to about 430°C. The average diameter and length of the nanowires is approximately 30 nm and 1.0 μ m, respectively; some of the nanowires grew as long as a few micrometers. Figure 2 shows SEM images of the as-synthesized nanostructures collected in the temperature region from about 450°C to about 550°C. In this temperature region, ZnO nanobelt-like structures were observed and they completely coated onto, and grew almost perpendicular to, the α -Fe₂O₃ nanobelts/nanowires. Energy dispersive X-ray spectroscopy (EDS) technique was used to determine the composition of the nanostructures grown on the α -Fe₂O₃ nanobelts/nanowires. Figure 3a shows a SEM image of a α -Fe₂O₃ nanowire coated with small ZnO nanosheets. Fig. 3b displays an EDS spectrum obtained from the small ZnO nanosheet (indicated by the white arrow in Fig. 3a), confirming that the as-synthesized ZnO nanowires/nanosheets contain a significant amount of Fe. Therefore, we have successfully synthesized Fe-doped ZnO nanowires and nanosheets. The detailed growth mechanisms and the compositional analyses of the as-synthesized ZnO/ α -Fe₂O₃ hetero-phase nano-architectures will be discussed [4].

References

- [1] W. Weiss et al., *Catal. Lett.* **52** (1998) 15.
 [2] J. Shi et al., *Mater. Lett.* **61** (2007) 5268.
 [3] Y. Fu et al., *Chem. Phys. Lett.* **379** (2003) 373.
 [4] This research was supported by the University of Missouri-St. Louis.

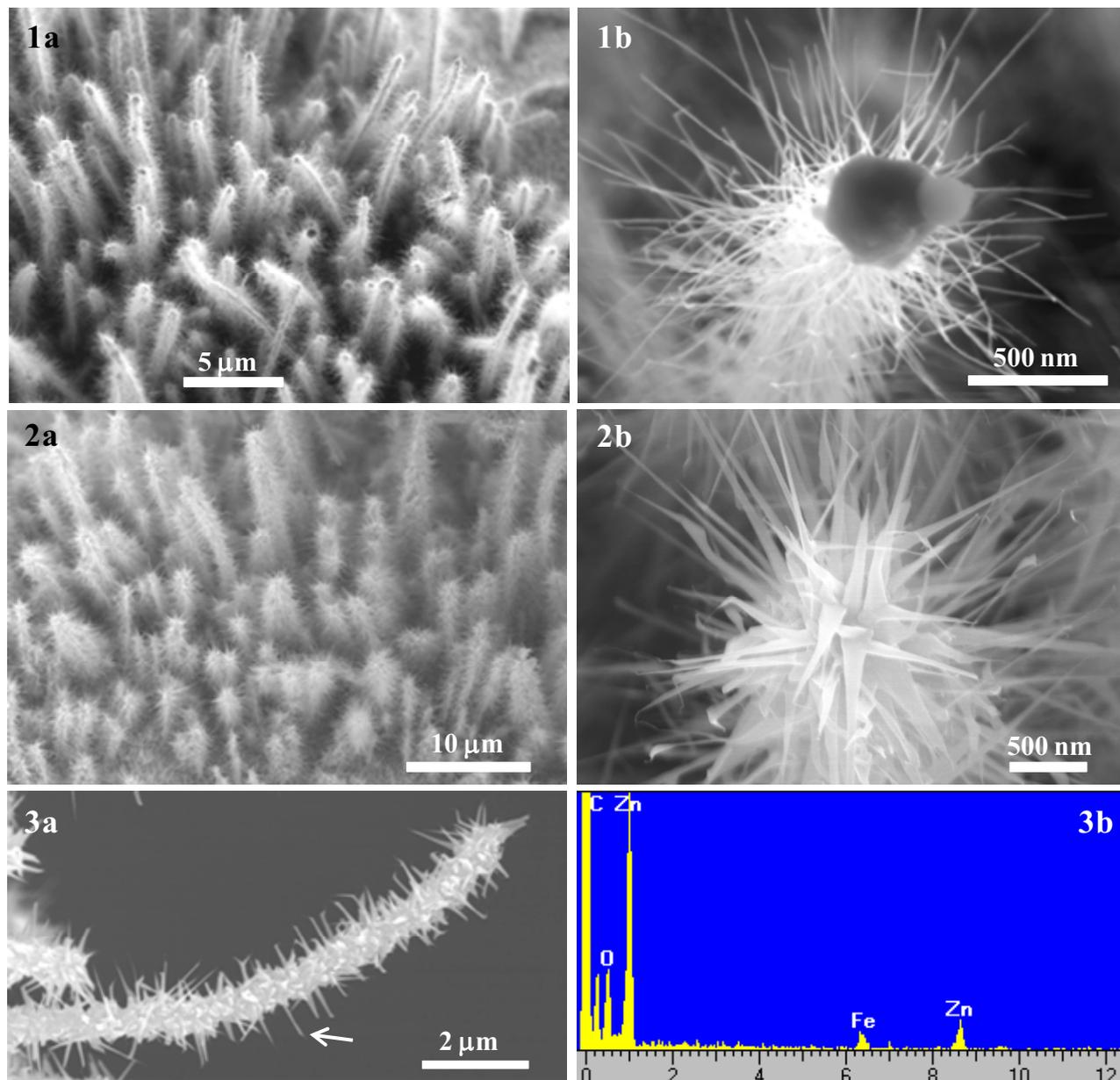


FIG. 1. Low (a) and high (b) magnification SEM images of ZnO nanowires growing almost perpendicular to the original big α -Fe₂O₃ nanowires in the lower temperature region.

FIG. 2. Low (a) and high (b) magnification SEM images of ZnO nanosheets growing almost perpendicular to the original big α -Fe₂O₃ nanowires in the higher temperature region.

FIG. 3 SEM image (a) and the EDS data taken from the small ZnO nanowire (indicated by the white arrow in 3a) showing presence of Fe in the ZnO nanowires.