

## ***In situ* Observation of Au/TiO<sub>2</sub> Catalysts in Reactant Gases by Environmental Transmission Electron Microscopy**

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Gold nanoparticles exhibit high catalytic activity when they are supported on selected metal oxides. Au/TiO<sub>2</sub> catalysts show great activity for CO oxidation even at low temperature [1]. Although it is well known that the catalytic activity of gold nanoparticles depends on the average size of them and support selection, the origin of the catalysis is still under debate [2]. Environmental transmission electron microscopy (ETEM) is useful to reveal the processes of reactions on nanoparticle catalysts [3, 4]. However, it is indispensable in ETEM observations to distinguish the effects of catalysis and those of electron irradiation. In this research, we observed Au/TiO<sub>2</sub> catalysts in reactant gases under electron irradiation with several different intensities.

An FEI Tecnai F20 transmission electron microscope equipped with an environmental cell was used for our observations. A differentially pumped system is adopted for the environmental cell, enabling us to perform high resolution observations in gases up to 2500 Pa (for N<sub>2</sub>). Au/TiO<sub>2</sub> catalysts were prepared by the deposition precipitation method and show catalytic activity for CO oxidation below room temperature. For ETEM observation, Au/TiO<sub>2</sub> catalysts were dispersed over a carbon microgrid on a copper mesh. The mesh was set on a specimen holder and transferred into the ETEM operated at 200 kV. Figure 2 shows typical TEM images of Au/TiO<sub>2</sub> catalysts.

Figure 3 shows sequential ETEM images of a Au/TiO<sub>2</sub> catalyst in CO/Air (CO:O<sub>2</sub>:N<sub>2</sub>=1:21:78 vol%) at 100 Pa. Amorphous materials began to accumulate at the periphery of Au particles within one minute. Subsequently, some Au particles were lifted off and pillar-like structures were formed, and other Au particles were partially or completely encapsulated. Pillars and capsules often show the crystalline structures, which were identified as rutile-TiO<sub>2</sub> epitaxial to the support rutile-TiO<sub>2</sub>. During the formation of pillars and capsules, Au particles were rotating and reconstructing the shape. To reveal the origin of these shape changes, we performed ETEM observations of Au/TiO<sub>2</sub> catalysts in high vacuum and pure O<sub>2</sub> gas with several different intensities of electron irradiation, considering the effects of electron irradiation on TiO<sub>2</sub> [5] and Au nanoparticles [6]. As a result, it is suggested that the formation of pillars and capsules are induced by two reactions: reduction of TiO<sub>2</sub> surface by electron irradiation and oxidation of Ti atoms or TiO<sub>2-x</sub> by O<sub>2</sub> gas or residual oxygen at the periphery of Au particles. These effects must be taken into consideration in ETEM observations of Au catalysts.

### References

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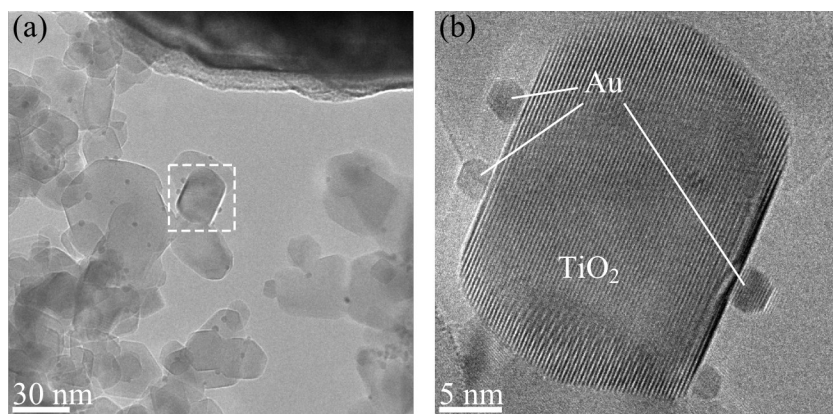


FIG. 1. (a) typical TEM image of Au/TiO<sub>2</sub> catalysts and (b) magnified image of (a).

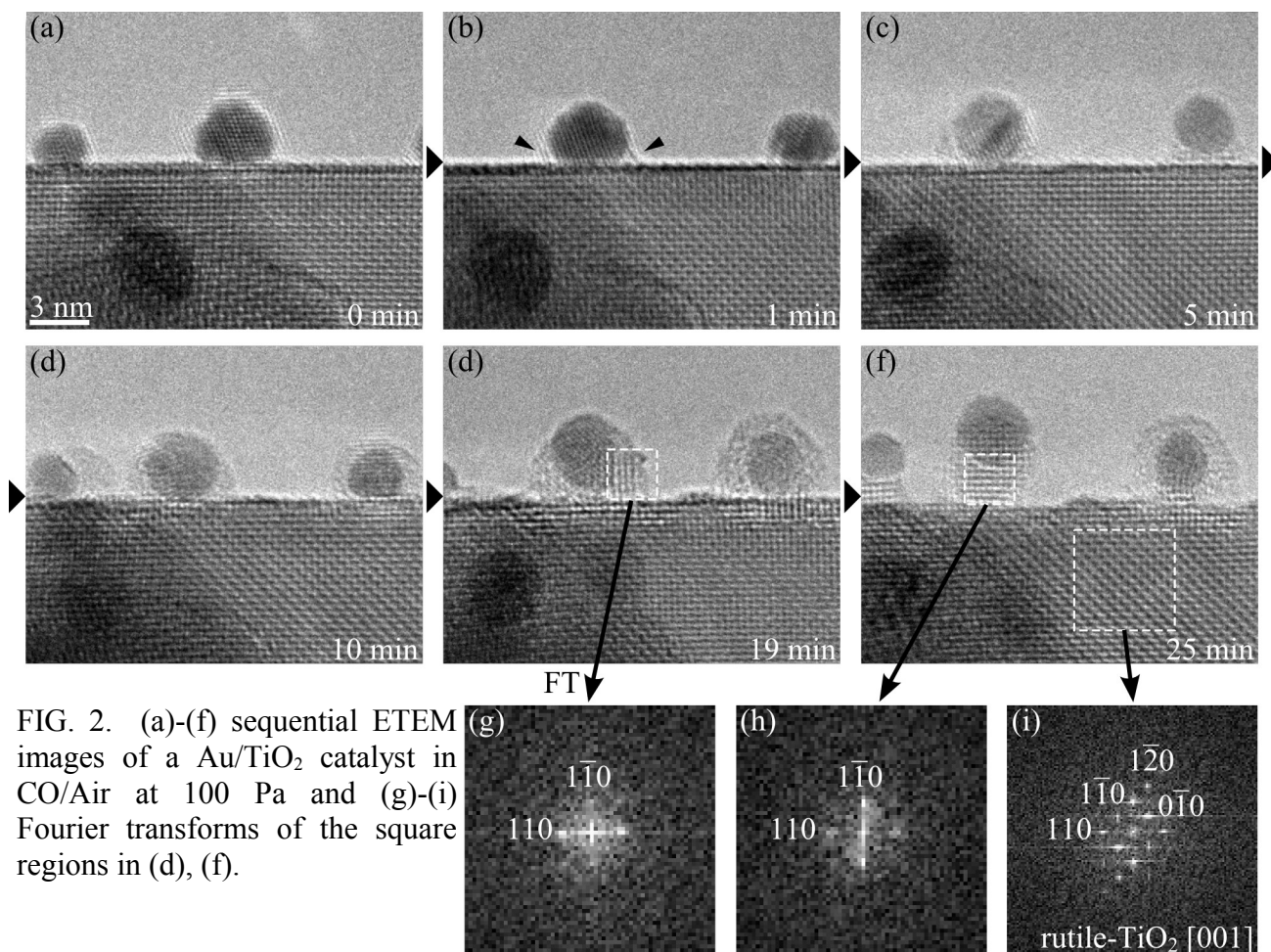


FIG. 2. (a)-(f) sequential ETEM images of a Au/TiO<sub>2</sub> catalyst in CO/Air at 100 Pa and (g)-(i) Fourier transforms of the square regions in (d), (f).