

## $\zeta$ -Factor Development and Quantification of a Boron Carbide and Silicon Hexaboride Diffusion Couple

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Boron-rich ceramics have been investigated as candidate materials for armor applications because of their high hardness and low density [1]. However, these materials are susceptible to amorphization under stress and exhibit low fracture resistance [2]. One strategy to reduce amorphization and improve fracture resistance is to incorporate Si-rich additives during processing. This strategy has been shown to be successful as dissolved Si in the B<sub>4</sub>C lattice mitigates amorphization [3], and Si-rich grain boundary complexions could alter bulk B<sub>4</sub>C fracture mechanisms similar to nanolayer complexions in silicon nitride [4]. While material properties are improved with Si-rich additives, there has been historically less progress to accurately quantify Si dissolution or grain boundary chemistry in boron-rich ceramics, especially using conventional analytical techniques such as X-ray energy dispersive spectroscopy (XEDS) or electron energy loss spectroscopy (EELS). Therefore, an analytical capability that can help link material performance to material chemistry in boron-based materials is needed. This work extended the  $\zeta$ -factor approach to develop a new capability to analyze boron-rich ceramics [5].

A diffusion couple with boron carbide (B<sub>4</sub>C) and silicon hexaboride (SiB<sub>6</sub>) was fabricated to produce Si-doped B<sub>4</sub>C, to measure the maximum Si solubility in the B<sub>4</sub>C lattice, and to quantify excess segregation levels of Si-doped B<sub>4</sub>C grain boundaries. Thin specimens were fabricated using a FEI Scios focused ion beam and XEDS analysis was completed using a JEOL ARM-200CF instrument equipped with a JEOL large-angle silicon drift detector.  $\zeta$ -factors of boron and carbon K $_{\alpha}$  lines were determined from high purity borides and carbides, respectively, and the  $\zeta$ -factors of other remaining K $_{\alpha}$  lines were determined from a NIST 2063a standard reference glass thin film. Spectrum images were acquired and grain boundary excess values were determined using a raster scan technique [6].

Figure 1 summarizes a spectrum image of a B<sub>4</sub>C / boron suboxide (B<sub>6</sub>O) interface. The B<sub>6</sub>O was a product of oxygen impurities during processing, but it was a convenient region of the specimen to validate the  $\zeta$ -factor method by comparing two different boron-rich compounds. Overall, the B<sub>4</sub>C/B<sub>6</sub>O interface is clearly resolvable and multiple second phases were located near the boundary. Furthermore, the primary second phase exhibited an internal two-phase mixture consisting of Si-O-rich and Al-Mg-O-rich regions and Mg and Al segregation to the B<sub>4</sub>C and B<sub>6</sub>O phase boundaries. SiB<sub>6</sub> and pure Si secondary phases were also observed. The boron concentrations of the B<sub>4</sub>C and B<sub>6</sub>O were  $87.3 \pm 2.0$  at.% and  $88.4 \pm 1.3$  at.% B, respectively. In addition, the Si concentration was measured to be  $1.5 \pm 0.1$  at.%, which was less than solubility limits found in the literature. All error values are reported as 95% certainty ( $2\sigma$ ).

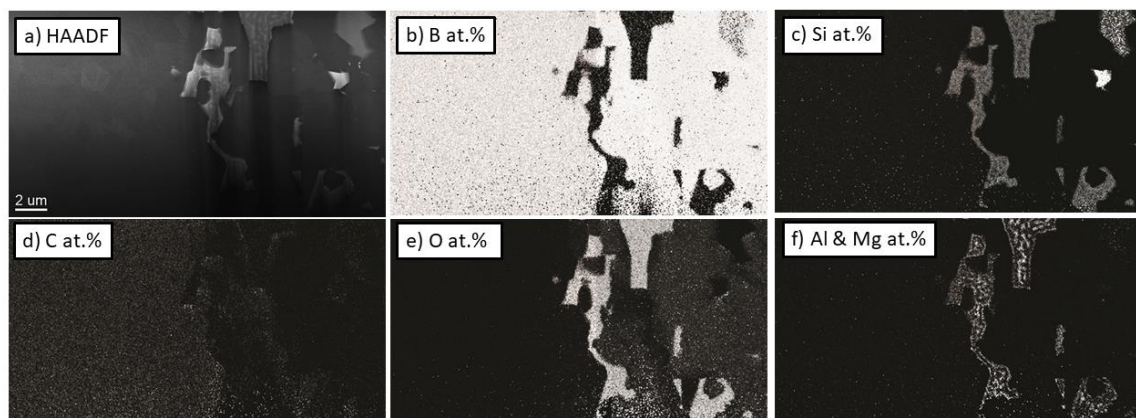
Grain boundary excess planar densities in B<sub>4</sub>C were quantified to determine the variability of grain boundary segregation. Figure 2 compares two grain boundaries: one boundary exhibits minimal segregation (a) and the other exhibits strong segregation (b). The excess planar densities of Si and Al in

the boundary exhibiting strong segregation were  $4.6 \pm 3.2$  and  $-0.02 \pm 0.15$  atoms/ $\text{\AA}^2$ , respectively, and the Si and Al excess planar densities in the boundary exhibiting minimal segregation were  $0.86 \pm 3.6$  and  $0.03 \pm 0.2$  atoms/ $\text{\AA}^2$ , respectively. In summary, the average excess planar densities were quantified and can be accurately measured using this methodology.

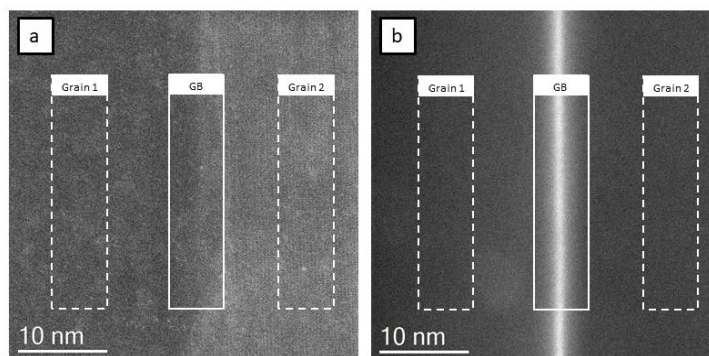
Overall, this  $\zeta$ -factor analytical framework to quantify boron-rich ceramics will provide a required insight to accelerate the materials design process. It can produce chemistry data with high accuracy and statistical confidence, and in combination with multiple grain boundary measurements, it can predict a grain boundary chemistry distribution for a given microstructure. Furthermore, these techniques can be utilized to investigate lattice and grain boundary concentrations in other light-element materials.

#### References:

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**Figure 1.** Figure 1a shows a reference high-angle annular dark field reference image, and Figures 1b–f show the boron, silicon, carbon, oxygen, aluminum, and magnesium composition maps.



**Figure 2.** Figures 2a and 2b show the grain boundaries that exhibited minimal and strong segregation. Each image is a high-angle annular dark field image and the three boxes visualize the raster scan method.